

Zn₂GeO₄ Nanowires As Efficient Electron Injection Material for Electroluminescent Devices

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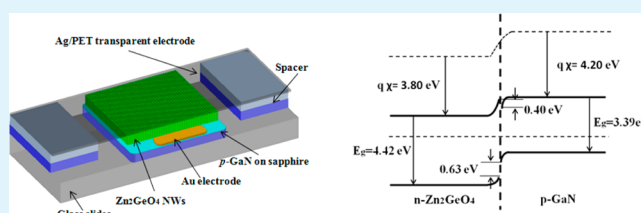
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Supporting Information

ABSTRACT: Pure phase Zn₂GeO₄ nanowires (NWs) were grown by the chemical vapor transport method on *p*-GaN:Mg/Al₂O₃ substrate. The as-grown Zn₂GeO₄ NWs exhibited *n*-type characteristic due to native defects and formed a *p*-*n* heterojunction with the *p*-GaN substrate. The unique energy level of Zn₂GeO₄ NWs promotes electron injection into GaN active region while suppressing hole injection into Zn₂GeO₄ NWs. The device exhibited an emission centered at 426 nm and a low turn-on voltage around 4 V. Zn₂GeO₄ NWs are first reported in this paper as promising electron transport and injection material for electroluminescent devices.

KEYWORDS: zinc germinate nanowires, *p*-GaN, light emitting diode, electron injection, electroluminescence, *p*-*n* heterojunction



1. INTRODUCTION

Nanowire electroluminescent devices have attracted extensive attention because of the many advantages of NWs, such as structural flexibility, high crystallinity, and nanoscale dimension. The unique one-dimensional structure of NWs provides a spontaneous mechanism for strain relaxation and lattice mismatch accommodation,¹ making it viable to integrate other functional materials onto III-V optically active semiconductors with low defects. This advantage helps to increase carrier injection efficiency and contribute in tackling the challenging problem of efficiency droop in light-emitting diode (LED), which is partially related to the high dislocation density caused by lattice-mismatch epitaxy.² Moreover, NWs provide direct pathways for photon transport, improved photon extraction is expected in the one-dimensional nanostructure. Apart from being used as the light emitting active elements, nanowires are also good candidates for carrier injection materials. The capability to grow high-crystal-quality nanowires on the substrates with less lattice match requirement³⁻⁵ enables a large range of materials selection for carrier transport and injection, which opens up numerous freedoms to engineer band structures of the *p*-*n* junctions and improve the LEDs performance.

Zn₂GeO₄ is a ternary functional oxide with many unique and interesting properties, such as negative thermal expansion below room temperature,⁶ white luminescence due to native defects,⁷ and high photon response for ultraviolet light detection.⁸ Earlier optoelectronic studies mainly focused on its properties in high field electroluminescence because of some of its advantages: extreme stability under high electrons irradiation, presence of native defects that act as effective emission centers and large bandgap that can avoid self-absorption of emitted light.^{7,9-11} However, it is challenging to

explore this material for application in LED devices in the bulk structure due to its relatively high resistivity (1×10^{11} Ohm cm based on bulk structure¹²) and large lattice mismatch with optically active III-V materials. In this work, we synthesized Zn₂GeO₄ single-crystal nanowires on III-V substrate and fabricate nanowire LED device using Zn₂GeO₄ nanowires as an efficient charge injection layer based on its unique band structure, efficient charge transport and tolerance to lattice mismatch. It was demonstrated that LED devices could be successfully fabricated by integrating Zn₂GeO₄ NWs with *p*-GaN substrate. The electrical and optical properties of Zn₂GeO₄ NWs in injection electroluminescent devices were elucidated.

2. EXPERIMENTAL SECTION

The Zn₂GeO₄ NWs were synthesized by a chemical vapor transport method.¹³ ZnO, GeO₂, and carbon powders were mixed with molar ratio of 2:1:3. The mixed powder was put into a 30 mm diameter quartz tube and inserted into the furnace, which was heated up to 1000 °C. Ar gas mixed with 10% O₂ was used as carrier gas. Si substrates coated with thin Au film (~10 nm) were used to collect NWs grown in the temperature region of 500–400 °C. The system pressure and oxygen concentration significantly affect morphology of the NWs (see Figure S1 in the Supporting Information). The growth conditions were chosen to achieve preferred NWs product based on the consideration of high electron injection efficiency and good carrier mobility. Detailed illustrations can be found in the Supporting Information. Morphology and

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structures of the NWs were characterized by field-emission scanning electron microscopy (SEM, JEOL 7600F), transmission electron microscopy (TEM, JEOL 2100), and X-ray diffraction (XRD, Shimadzu XRD-6000, Cu K α radiation).

Mg-doped GaN (0001) epitaxial film (3 μm in thickness) on Al $_2$ O $_3$ template was used as the *p*-type substrate. Zn $_2$ GeO $_4$ NWs were grown on the substrate by the method described above. The GaN substrate was partially covered and protected by a hard mask during the nanowire growth process. After the Zn $_2$ GeO $_4$ NW growth, 100 nm Au layer was sputtered onto the protected area and followed by a 500 $^\circ\text{C}$ rapid thermal annealing for 3 min in air to form Ohmic contact. The prepared sample was placed onto a glass slide which served as a supporting substrate. Two clean GaN substrates with thin elastic films (paraffin film) on top were put on both sides of the sample to serve as spacers. Bulldog clips were used to press Ag/PET transparent electrode onto the spacers and assemble the device.^{14,15} By using this method, we can achieve good Ohmic contact between the top electrode and underneath NWs without complex processes of insulating layers deposition, plasma etching, and transparent electrode deposition.^{16,17}

3. RESULTS AND DISCUSSION

Top view of the Zn $_2$ GeO $_4$ NWs grown on *p*-GaN substrate is shown in Figure 1a. The lengths are typically $\sim 100 \mu\text{m}$ with a

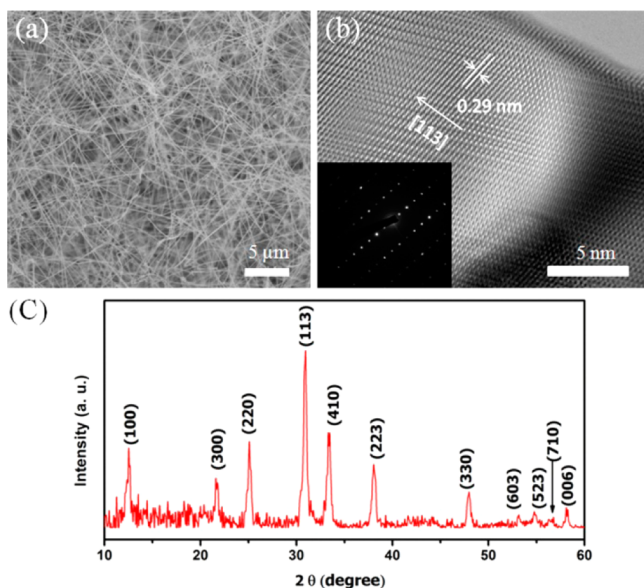


Figure 1. (a) SEM image of Zn $_2$ GeO $_4$ NWs. (b) High-resolution TEM image of the Zn $_2$ GeO $_4$ NWs, which have a growth direction of $\langle 113 \rangle$. Inset is the corresponding selective area diffraction patterns. (c) XRD pattern of the Zn $_2$ GeO $_4$ NWs.

growth time of 20 min. Figure 1b is a high-resolution TEM image of the Zn $_2$ GeO $_4$ NWs, showing single-crystalline structure of the NWs. Crystallinity of the NWs was further confirmed by the selective area electron diffraction (SAED) patterns, as shown in Figure 1b inset. Figure 1c is the XRD pattern of the Zn $_2$ GeO $_4$ NWs. All the peaks can be readily indexed to a pure rhombohedral crystal phase (JCPDS: 011-0687, $a = 14.231 \text{ \AA}$, $b = 9.53 \text{ \AA}$). The high-purity Zn $_2$ GeO $_4$ NWs eliminates the concern of impurity, such as ZnO NWs and GeO $_2$ NWs, which may affect performance of the devices.

Room-temperature photoluminescence (PL) of the Zn $_2$ GeO $_4$ NWs on Si substrate and GaN substrate are presented in Figures 2a, b. The Zn $_2$ GeO $_4$ NWs have a broad emission from

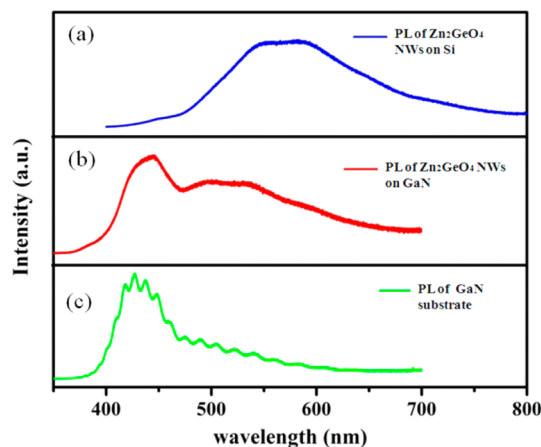


Figure 2. (a) Room-temperature PL spectrum of Zn $_2$ GeO $_4$ NWs on Si substrate. (b) PL spectrum of Zn $_2$ GeO $_4$ NWs on GaN substrate. (c) PL spectrum of pristine GaN substrate.

500 to 700 nm, which was ascribed to the donor–acceptor recombination from native defects.⁷ Since the excitation energy (3.81 eV) was relatively low compared to the bandgap of Zn $_2$ GeO $_4$ (4.42 eV), the band edge emission behavior of the NWs could not be shown. PL of the Zn $_2$ GeO $_4$ NWs on GaN substrate was a combination of the narrow emission centered at $\sim 440 \text{ nm}$ and the broad emission from native defects of Zn $_2$ GeO $_4$ NWs. The emission centered at $\sim 440 \text{ nm}$ came from the GaN substrate in which transits were from conduction band or shallow donors to the deep-level Mg dopants.¹⁸ Figure 2c shows the PL from pristine GaN substrate, which confirms the 440 nm emission from GaN. The multiplex structure in the spectrum was attributed to light interference in the GaN film.¹⁹

Figure 3a is a schematic of the Zn $_2$ GeO $_4$ NW/*p*-GaN heterojunction device. Silver nanoparticles on PET substrate²⁰ (Ag/PET, sheet resistivity of 2 ohm/sq, transparency of $\sim 65\%$ from 900 to 370 nm) was used as the top transparent electrode by mechanical press onto the Zn $_2$ GeO $_4$ NWs. Reliable current feeding and light extraction were achieved by this device configuration. Current–voltage (*I*–*V*) characteristic of the device was measured by a Keithley 4200 semiconductor analyzer. The device exhibited strong rectifying behavior with remarkable small reverse current, as presented in Figure 2b. Due to oxygen vacancy and zinc interstitial, as donors, the NWs behaved as intrinsic *n*-type materials.²¹ The *p*–*n* heterojunction was formed between the Zn $_2$ GeO $_4$ NWs and *p*-GaN substrate. Electrical properties of the device were further investigated by measuring the *I*–*V* characteristics under varied cryogenic temperature (see Figure S2 in the Supporting Information). It shows that Zn $_2$ GeO $_4$ NWs had a carrier freeze-out temperature at around 150 K. Detailed results and discussions are provided in the Supporting Information.

Strong blue emission from the device could be observed under forward bias, as shown in the inset of Figure 3c. No emission was detected under reverse bias. The emission spectra centered at 426 nm with a full width at half-maximum (fwhm) of about 60 nm, as shown in Figure 3c. The electroluminescence process can be explained by energy band diagram of the heterojunction in Figure 3d. The conduction band edge

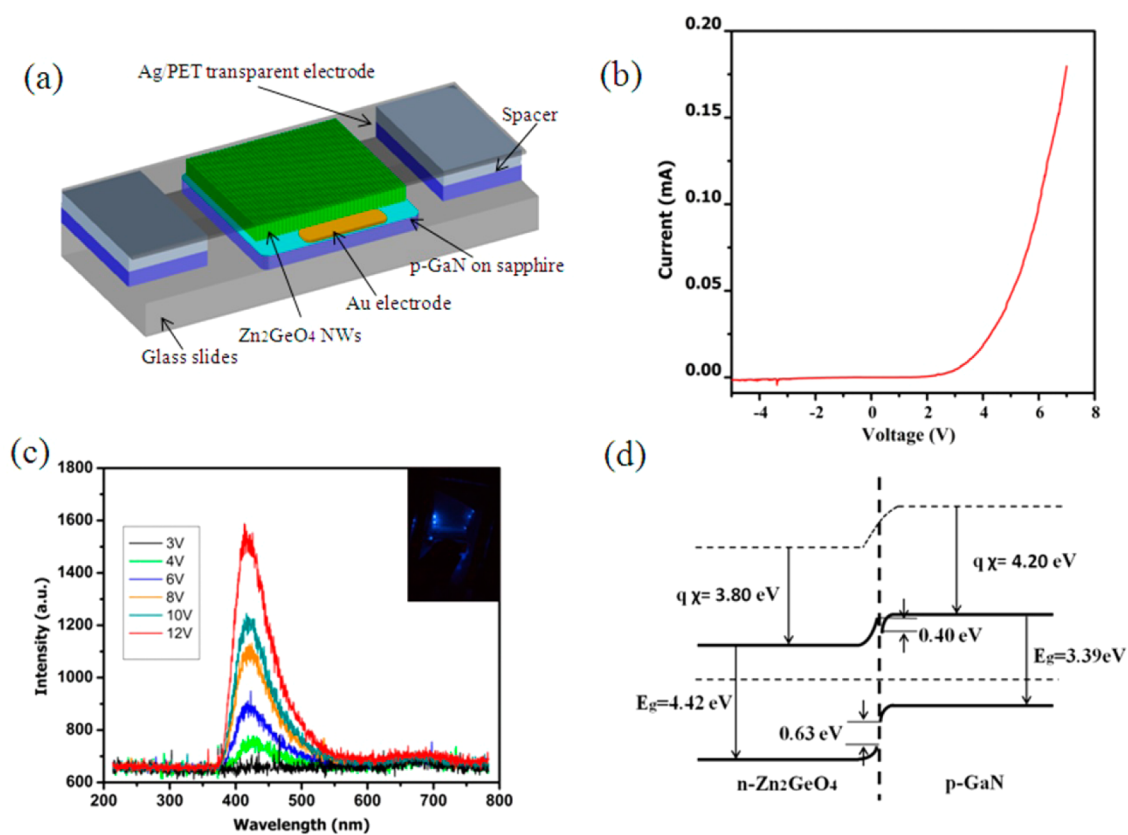


Figure 3. (a) Schematic of the Zn₂GeO₄ NWs/GaN heterojunction device. (b) Current–voltage curve of the LED device. (c) Electroluminescence spectra of synthesized device, inset is a lighted LED device under 10 V forward bias. (d) Energy band diagram of the heterojunction.

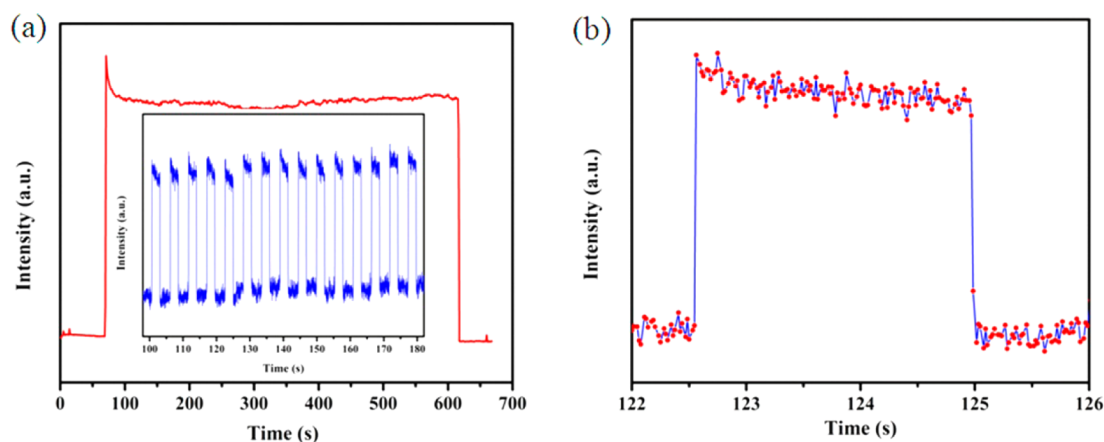


Figure 4. (a) EL stability test of the device with an operation time of 10 min under 8 V forward bias. Inset is the EL time respond of the device under rectangle pulse function. (b) Enlarged image for response of the device under a single pulse.

of Zn₂GeO₄ is located at around -3.80 eV (vs AVS-absolute vacuum state),²² 0.4 eV higher than the conduction band of GaN (-4.20 eV vs AVS).²³ Electrons can be easily injected into GaN in this energy band structure.

It was reported that a Ga₂O₃ film formed on GaN substrates before the NWs growth in a chemical vapor deposition process. The Ga₂O₃ barrier reduced electron injection efficiency under forward bias and largely increased the turn on voltage of the LED device to +11.5 V.²⁴ The large turn-on voltage was not observed in our devices which have a low turn-on voltage at around 4 V. This suggests that the Zn₂GeO₄ NWs should be in direct contact with GaN thin film and high quality heterojunction was formed in the device. The ease of electron

injection greatly enhanced the minority carrier concentrations in the GaN active regions under forward bias, which increased the radiative recombination rate while the nonradiative recombination rate may be suppressed due to saturation of deep level traps.²⁵ There is a 0.63 eV energy barrier ($\Delta E_v = 0.63$ eV) between the valence bands of Zn₂GeO₄ and GaN. Hole injection into the conduction band of Zn₂GeO₄ NWs is largely suppressed under the high energy barrier. Because of the low injection efficiency, these minority carriers more likely relaxed through nonradiative recombination process. Consequently, no characteristic emission from Zn₂GeO₄ NWs was observed in the EL spectra. No emission was detected from the device with forward voltage below 3 V. When the bias was

increased to 4 V, slightly above the bandgap of GaN, high electrons injection took place, resulting in emission from the GaN centered at 426 nm. The low turn-on voltage was attributed to good Ohmic contact in the device and high quality interface at the nanocontact of Zn_2GeO_4 NWs and GaN film.

Dynamic switching experiments were also performed to investigate stability and time response of the devices, as presented in Figure 4. Emission intensity of the device was quite stable during the test with 8 V forward bias. In order to test the time response behaviors, a rectangular pulse function (pulse width of 2.5 s, pulse period of 5 s and pulse voltage of 8 V) was applied to the device. As shown in Figure 4a inset, the device showed quick and repeatable response to the pulse function. The response time was below 16 ms, which was limited by the acquisition time of our spectrometer, as presented in one enlarged cycle in Figure 4b.

4. CONCLUSION

In conclusion, we demonstrate that the wide-bandgap intrinsic n-type ternary material, Zn_2GeO_4 NWs, can be used to form high quality p–n heterojunction with p-type GaN for electroluminescent applications. The conduction band offset (from Zn_2GeO_4 to GaN) ΔE_c is -0.4 eV, whereas the valence band offset ΔE_v is 0.63 eV. The p–n heterojunction greatly facilitates electron injection into p-GaN active materials while suppressing hole injection into Zn_2GeO_4 . As a result, main radiative recombination process was confined in GaN active region, which has high internal quantum efficiency. The large bandgap of Zn_2GeO_4 NWs also reduces self-absorption of the emission and increases light extraction efficiency. Therefore, the ternary oxide NWs are promising materials for electroluminescent devices as electron injection layers.

■ ASSOCIATED CONTENT

Supporting Information

Additional figures and details (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) Yang, P.; Yan, R.; Fardy, M. *Nano Lett.* **2010**, *10*, 1529–1536.
- (2) Guo, W.; Zhang, M.; Banerjee, A.; Bhattacharya, P. *Nano Lett.* **2010**, *10*, 3355–3359.
- (3) Martensson, T.; Svensson, C. P. T.; Wacaser, B. A.; Larsson, M. W.; Seifert, W.; Deppert, K.; Gustafsson, A.; Wallenberg, L. R.; Samuelson, L. *Nano Lett.* **2004**, *4*, 1987–1990.
- (4) Kuykendall, T.; Ulrich, P.; Aloni, S.; Yang, P. *Nat. Mater.* **2007**, *6*, 951–956.

- (5) Calarco, R.; Meijers, R. J.; Debnath, R. K.; Stoica, T.; Sutter, E.; Lüth, H. *Nano Lett.* **2007**, *7*, 2248–2251.
- (6) Stevens, R.; Woodfield, B. F.; Boerio-Goates, J.; Crawford, M. K. *J. Chem. Thermodyn.* **2004**, *36*, 349–357.
- (7) Liu, Z.; Jing, X.; Wang, L. *J. Electrochem. Soc.* **2007**, *154*, H500–H506.
- (8) Yan, C.; Singh, N.; Lee, P. S. *Appl. Phys. Lett.* **2010**, *96*, 053108.
- (9) Anoop, G.; Mini Krishna, K.; Jayaraj, M. K. *J. Alloys Compd.* **2009**, *468*, 512–515.
- (10) Anoop, G.; Krishna, K. M.; Jayaraj, M. K. *J. Electrochem. Soc.* **2008**, *155*, J7–J10.
- (11) Lewis, J. S.; Holloway, P. H. *J. Electrochem. Soc.* **2000**, *147*, 3148–3150.
- (12) Struck, C. W.; Mishra, K. C.; Bartolo, B. D. *Physics and Chemistry of Luminescent Materials*; The Electrochemical Society: Pennington, NJ, 1999; p 142.
- (13) Yan, C.; Lee, P. S. *J. Phys. Chem. C* **2009**, *113*, 14135–14139.
- (14) Lupan, O.; Pauporte, T.; Viana, B. *Adv. Mater.* **2010**, *22*, 3298–3302.
- (15) Lupan, O.; Pauporté, T.; Viana, B. *J. Phys. Chem. C* **2010**, *114*, 14781–14785.
- (16) Xu, S.; Xu, C.; Liu, Y.; Hu, Y.; Yang, R.; Yang, Q.; Ryou, J. H.; Kim, H. J.; Lochner, Z.; Choi, S.; Dupuis, R.; Wang, Z. L. *Adv. Mater.* **2010**, *22*, 4749–4753.
- (17) Zhang, X.-M.; Lu, M.-Y.; Zhang, Y.; Chen, L.-J.; Wang, Z. L. *Adv. Mater.* **2009**, *21*, 2767–2770.
- (18) Alivov, Y. I.; Van Nostrand, J. E.; Look, D. C.; Chukichev, M. V.; Ataev, B. M. *Appl. Phys. Lett.* **2003**, *83*, 2943.
- (19) Dai, J.; Xu, C. X.; Sun, X. W. *Adv. Mater.* **2011**, *23*, 4115–4119.
- (20) Layani, M.; Magdassi, S. *J. Mater. Chem.* **2011**, *21*, 15378–15382.
- (21) Liu, Z.; Liang, B.; Chen, G.; Yu, G.; Xie, Z.; Gao, L.; Chen, D.; Shen, G. *J. Mater. Chem. C* **2013**, *1*, 131–137.
- (22) Liu, Q.; Zhou, Y.; Kou, J.; Chen, X.; Tian, Z.; Gao, J.; Yan, S.; Zou, Z. *J. Am. Chem. Soc.* **2010**, *132*, 14385–14387.
- (23) Jeong, M.-C.; Oh, B.-Y.; Ham, M.-H.; Myoung, J.-M. *Appl. Phys. Lett.* **2006**, *88*, 202105.
- (24) Yang, H. Y.; Yu, S. F.; Liang, H. K.; Lau, S. P.; Pramana, S. S.; Ferraris, C.; Cheng, C. W.; Fan, H. J. *ACS Appl. Mater. Interfaces* **2010**, *2*, 1191–1194.
- (25) Mueller, G. *Electroluminescence I*; Academic Press: San Diego, CA, 2000; p 15.